

tent \bar{U}_1 and \bar{U}_4 respectively are obtained as a function of the dimensionless drying time N_{F_0} and the percentage differences $[(\bar{U}_1 - \bar{U}_4)/\bar{U}_1]$ 100 per cent are plotted as a function of N_{F_0} with $N_{L_{u_1}}$ as parameter where "V" indicates the beginning of the falling rate period.

It is evident in Fig. 4 that the "intermittent" procedure improves the efficiency even during the first interval of the falling rate period. This is particularly true for materials exhibiting small values of $N_{L_{u_1}}$, i.e. for porous solids which are difficult to dry. It has been shown previously [1] that the moisture content profiles corresponding to small $N_{L_{u_1}}$ has a much steeper profile towards the surface. Therefore, at the end of each constant rate period, the remaining average moisture content, \bar{U} , is larger for small values of $N_{L_{u_1}}$. Consequently, the amount that can be removed during additional constant rate period is also larger. However, it should also point out that such a steep moisture content profile toward the surface may also be resulted by intensified external transfer processes or by additional means of heat supply. Therefore, one may conclude that such an "intermittent" process will be especially feasible for any drying processes whereby the duration of the constant rate

period is severely shortened either by intensifying the external transfer processes or by additional heat supply to the drying material. Furthermore, the general agreement between the results of experimental observation reported in literature and the prediction of the parameter study presented here, may be looked upon as a partial verification of the theoretical model for the drying process.

REFERENCES

1. D. Berger and D. C. T. Pei, Drying of hygroscopic capillary porous solids—a theoretical approach, *Int. J. Heat Mass Transfer* **16**, 283–302 (1973).
2. T. K. Sherwood, The drying of solids—I, II, *Ind. Engng Chem.* **21**, 12–16, 976–980 (1929).
3. P. Gorling, Drying behavior of vegetable substances, *ForschHft. Ver. Dt. Ing.* 458 (1956).
4. P. Gorling, Research results and research trends in drying technology, *Chem.-Ingr.-Tech.* **29**, 170–176 (1957).
5. W. Schicketany, Vergleich des Trocknungsverhaltens von Schuttungen Kapillar Poroser Materialien, *Chem.-Ingr.-Tech.* **43**, 245–251 (1971).

Int. J. Heat Mass Transfer. Vol. 18, pp. 709–710. Pergamon Press 1975. Printed in Great Britain

HEAT-TRANSFER REGIMES IN VERTICAL, PLANE-WALLED, AIR-FILLED CAVITIES

M. DIXON and S. D. PROBERT

School of Mechanical Engineering, Cranfield Institute of Technology, Cranfield, Bedford MK43 0AL

(Received 18 August 1974)

NOMENCLATURE

C_p ,	specific heat of air at atmospheric pressure;
d ,	thickness of the air layer;
g ,	local acceleration due to gravity;
Gr ,	Grashof number based on d ($\equiv g/\beta\Delta T d^3 \rho^2/\mu^2$);
h ,	height of the fluid layer;
hd ,	aspect ratio;
k ,	thermal conductivity of air;
Pr ,	Prandtl number ($\equiv \frac{\mu C_p}{k}$);
Ra ,	Rayleigh number ($\equiv Gr \cdot Pr$).

Greek symbols

β ,	coefficient of thermal expansion of the air;
ρ ,	density of the air at atmospheric pressure;
μ ,	dynamic viscosity of the air;
ΔT ,	temperature difference across the air layer of thickness d .

INTRODUCTION

FREE CONVECTIVE flow phenomena in plane, enclosed, vertical air layers at atmospheric pressure are of a complex nature. Three regimes of heat transfer can be distinguished within vertical cavities for the laminar flow region of Grashof number and these have been termed the conduction, the transition and the boundary-layer regimes respectively.

In the conduction regime heat is transferred predominantly by gaseous conduction and the temperature gradients are linear except near the top and bottom of the cavities. In the transition regime convective flow becomes significant

and the temperature profiles across the air layer are no longer linear. However, the boundary layers on the hot and cold walls merge with one another and gaseous conduction remains the dominant heat-transfer mechanism. The boundary layer regime is characterized by separate thermal boundary layers on the hot and relatively cold walls and the temperature profiles exhibit steep gradients at the walls with a zero or inverted slope within the core region. Consequently convection rather than gaseous conduction is the predominant mechanism of heat transfer.

LIMITS OF THE REGIMES

Eckert and Carlson [1] proposed that the delineation of the three regimes could be obtained from relationships of the form:

$$Gr = f(h/d). \quad (1)$$

A limit for the conduction regime has been derived theoretically by Batchelor [2] to be:

$$Gr = 695(h/d). \quad (2)$$

It is debatable, however, whether the limits of the flow regimes are dependent upon the aspect ratio (h/d), and no conclusive evidence has yet been offered to support this contention. For fluids of various Prandtl numbers (> 1), MacGregor and Emery [3] found that the flow regimes were characterised by the Rayleigh number. Moreover, it has been stated by de Graaf and Van der Held [4] that the aspect ratio has only a minor effect in controlling the flow conditions. Hence Brooks and Probert [5, 6] conducted an interferometric investigation on air layer widths of 0.635, 1.27

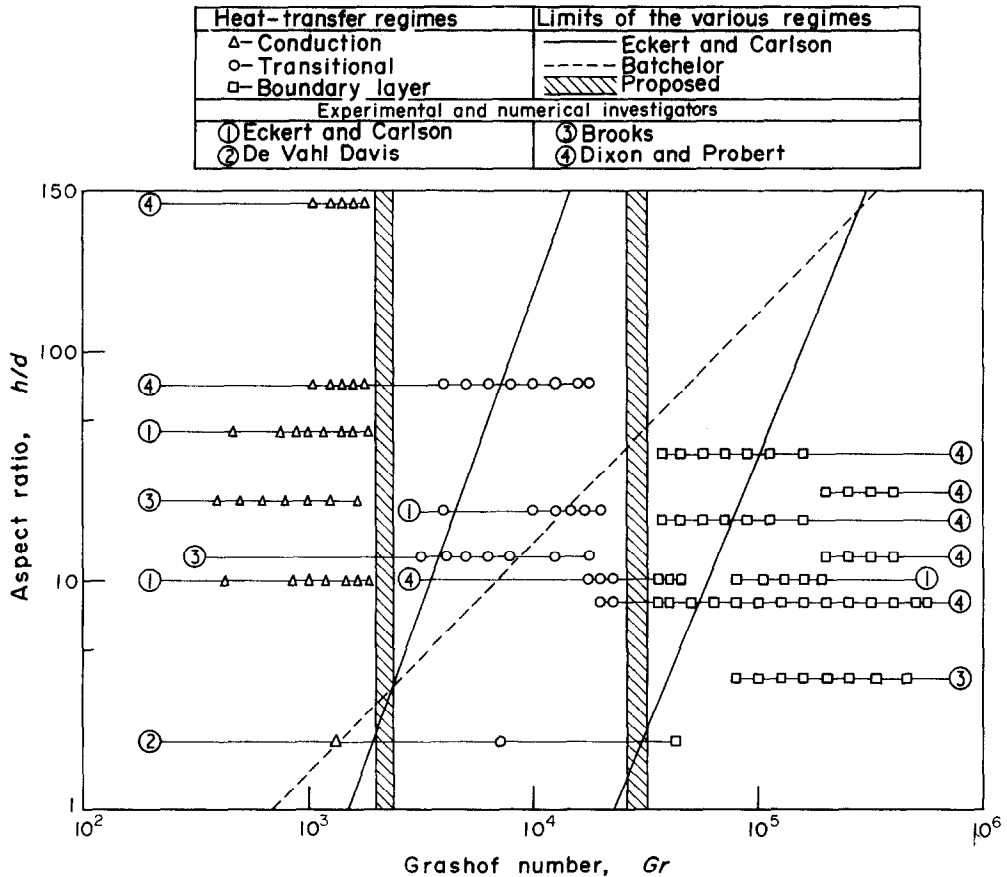


FIG. 1. Limits for the heat-transfer regimes for enclosed, plane walled, vertical layers.

and 3.81 cm with aspect ratios of 22.4, 11.2 and 3.73 respectively. The limits of the conduction and transition regimes for this study occurred at $Gr = 2.2(\pm 0.2) \times 10^3$ and $Gr = 2.9(\pm 0.3) \times 10^4$ respectively. Figure 1 illustrates the various limits for all the available results.

CONCLUSIONS

It can be seen from Fig. 1 that the limits of the regimes of heat transfer within vertical air layers can be identified solely in terms of the Grashof number and are independent of h . This implies that the cavity width d is the only geometric factor which determines the regimes, and therefore, it would appear that the Nusselt vs Grashof number relations should also be correlated on the width d rather than with respect to the aspect ratio (h/d).

Acknowledgement—The authors wish to thank the Science Research Council for support of this project.

REFERENCES

1. E. R. G. Eckert and W. O. Carlson, Natural convection in an air layer enclosed between two vertical planes with different temperatures, *Int. J. Heat Mass Transfer* **2**, 106 (1961).
2. G. K. Batchelor, Heat transfer by free convection across a closed cavity between vertical boundaries at different temperatures, *Q. Appl. Math.* **12**, 209 (1954).
3. R. K. MacGregor and A. F. Emery, Free convection through vertical plane layers—moderate and high Prandtl number fluids, *Trans. Am. Soc. Mech. Engrs* **91C**, 391 (1969).
4. J. G. A. de Graaf and E. F. M. Van der Held, The relation between the heat transfer and the convection phenomena in enclosed plane layers, *Appl. Scient. Res.* **3A**, 393 (1953).
5. R. G. Brooks, An interferometric study of heat transfer through gases, Ph.D. Thesis, University of Wales (1970).
6. R. G. Brooks and S. D. Probert, Heat transfer between parallel walls: an interferometric investigation, *J. Mech. Engng Sci.* **14**(2), 107–127 (1972).